

LAPSHIN, Fedor Alekseyevich; KOMAROV, Sergey Georgiyevich; BOCHARNIKOVA,
K.N., inzhener, redaktor; YUDZON, D.M., tekhnicheskiy redaktor.

[Railroad cars] Vagonnoe khoziaistvo. Moskva, Gos.transp.zhel-dor.
izd-vo, 1955. 190 p. (MLRA 8:9)
(Railroads--Cars)

KOMAROV, S.G.

FRANTSIV, Andrey Nikolayevich; KOMAROV, S.G., red.; VENINA, G.P., tekhn.red.

[Mechanist's handbook on repair of freight cars] Posobie slesariu
po remontu gruzovykh vagonov v pozdakh. Moskva, Gos. transp. shel-
dor. izd-vo, 1958. 190 p.
(MIRA 11:5)
(Railroads--Freight cars--Maintenance and repair)

KOMAROV, S.G.; SAMOKHVALOV, S.F.; BELEVNTSEV, N.V.; BOMBARDIROV, P.P.;
VERINA, A.A.; BLIZNYUK, V.F.; LADYGIN, V.I.; PAROV, A.N.; VASIL'YEV,
I.P.; BRODOVICH, N.B.; RABINOV, A.M.; ALEKSEYEV, V.D.; YEGOROV,
V.A., inzh., red.; ARSHINOV, I.M., inzh., red.; VERINA, G.P., tekhn. red.

[Handbook on the repair of freight cars] Spravochnik po remontu
gruzovykh vagonov. Moskva, Gos. transp. zhel-dor. izd-vo, 1958. 503 p.
(MIRA 11:12)

(Railroads--Freight cars--Maintenance and repair)

FRANTSEV, Andrey Nikolayevich; KOMAROV, S.G., red.; VERINA, G.P.,
tekhn.red.

[Mechanic's manual for the maintenance of freight cars in
operation] Posobie slesariu po remontu gruzovykh vagonov v
poezdakh. Izd.2., perer. i dop. Moskva, Gos.transp.zhal-dor.
izd-vo, 1959. 235 p. (MIRA 12:12)
(Railroads--Freight cars--Maintenance and repair)

KOMAROV, S.G.; KITOV, A.N., inzh.; DOROFEEV, V.G.; SHEREMET'YEV,
M.A.; FOMIN, A.A.; KOSAREV, A.A.; SARANTSEV, Yu.S., red.;
VERINA, G.P., tekhn.red.

[Handbook for the repair of passenger cars] Spravochnik po
remontu passazhirskikh vagonov. Moskva, Vses.izdatel'sko-
poligr. ob"edinenie M-va putei soobshcheniya, 1960. 631 p.

(MIRA 13:6)
(Railroads--Passenger cars--Maintenance and repair)

PHASE I BOOK EXPLOITATION

SCV/5872

Komarov, S. G. Doctor of Technical Sciences, ed.

Spravochnik geofizika, v chetyrekh tomakh. t. 2: Geofizicheskiye metody issledovaniya skvazhin (The Geophysicist's Handbook, in Four Volumes. v. 2: Geophysical Methods in the Exploration of Wells) Moscow, Gostoptekhizdat, 1961. 760 p. Errata slip inserted. 6130 copies printed.

Editorial Board: V. V. Fedynskiy, Chairman, V. N. Dakhnov, V. G. Vasil'yev, Ye. N. Kalenov, S. G. Komarov, M. K. Polshkov, L. A. Ryabinkin; Executive Ed.: Ye. G. Pershina; Tech. Ed.: E. A. Mukhina.

PURPOSE: The book is intended for scientific workers in the field of industrial geophysics.

COVERAGE: This volume of the four-volume Geophysicist's Handbook series deals with the geophysical exploration wells. It contains data on various types of logging: electrical, radioactivity, gas, induction, sonic, magnetic, etc. Problems of ground selection, perforation, and well shooting are analyzed. The theory behind the various methods is briefly outlined. The apparatus and equipment used in various industrial geophysical explorations are described, and the

Card 1/5

APPROVED FOR RELEASE: 06/13/2000

The Geophysicist's Handbook (Cont.)

CIA-RDP86-00513R000824110002-3"

SOV/5872

ways of interpreting the materials obtained are discussed. In addition, the handbook contains information on the organization of geophysical operations, safety measures, and general information on drilling. No personalities are mentioned. There are 341 references: 290 Soviet, 46 English, and 5 French.

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AVAILABLE: Library of Congress	
SUBJECT: Geophysics	

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MM/dwm/gmp
2-2-62

KASHCHEYEV, Nikolay Tarasovich; VALETOV, Aleksandr Ivanovich; KOMAROV,
Sergey Georgiyevich; POGORELYY, B.G., inzh., retsenzent;
SARANTSEV, Yu.S., inzh., red.; BOBROVA, Ye.N., tekhn. red.

[Manual on the structures and equipment of railroad car maintenance
and repair depots] Spravochnik po sooruzheniyam i oborudovaniiu
vagonnogo khoziaistva. Moskva, Transzheldorizdat, 1962. 423 p.
(MIRA 15:6)

(Railroads-- Cars). (Railroads--Repair shops)

KOMAROV, S.G.; SHMAROVA, V.P.

Membrane potential of clay. Prikl. geofiz. no.31:288-293 '61.
(Clay--Electric properties) (MIRA 15:3)

ANDREYEV, Mikhail Grigor'yevich; SMOL'YANINOVA, Aleksandra Mitrofanovna;
KOLEDENKOV, Sergey Semenovich; KOMAROV, Sergey Georgiyevich;
SHMANTSAR', D.N., retsenzent; DOROFEEVA, A.I., retsenzent;
PESKOVA, L.N., red.; VOROTNIKOVA, L.F., tekhn. red.

[Planning, business accounting and analysis of the administrative
operations of a railroad car depot] Planirovanie, khozraschet i
analiz khoziaistvennoi deiatel'nosti vagonnogo depo. Moskva,
Transzheldorizdat, 1962. 149 p. (MIRA 15:12)
(Railroads--Finance)

KOMAROV, S.I.

Technical conference on industrial water supply to metallurgical
and by-product coking plants. Prom.energ. 15 no.3:51
Mr '60. (MIRA 13:6)

(Water--Distribution)

KOMAROV, S.I.

Competition for the best suggestion on the economy of fuel in ferrous metallurgy plants. Prom.energ. 16 no.5:12-14 My '61.

(MIRA 14:7)

(Fuel) (Metallurgical plants)

L 8132-66

ACC NR: AP5025072

SOURCE CODE: UR/0286/65/000/016/0130/0131

B

AUTHOR: Komarov, S. K.

ORG: none

TITLE: Device that prevents winding of net ropes on ship propeller shafts. Class 65,
No. 174084

SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 16, 1965, 130-131

TOPIC TAGS: ship screw, marine propeller

ABSTRACT: This Author Certificate presents a device to prevent winding of net ropes, cables, etc onto ship propeller shafts. The device includes a shaft cover placed between the stern post and the ship propeller (see Fig. 1). To provide reliable protection by forming a directed flow which will turn aside the ropes, the cover is made in the form of two cone-shaped rings which are axially joined at their widest diameter and connected to the divider which in turn is attached to the stern post in the diametral plane of the ship.

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UDC: 629.1.037.4:621-783.632.1

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ACC NR: AP5025072

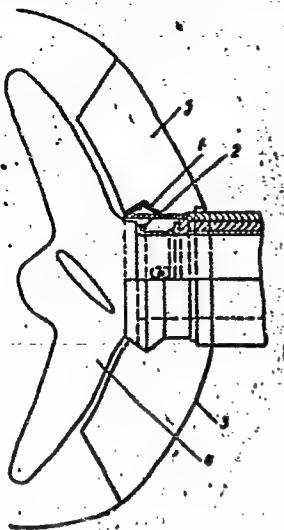


Fig. 1. 1- cover; 2- propeller shaft;
3- stern post; 4- ship screw;
5- separator

Orig. art. has: 1 figure.

SUB CODE: PR, GO, IE/ SUBM DATE: 17Mar64

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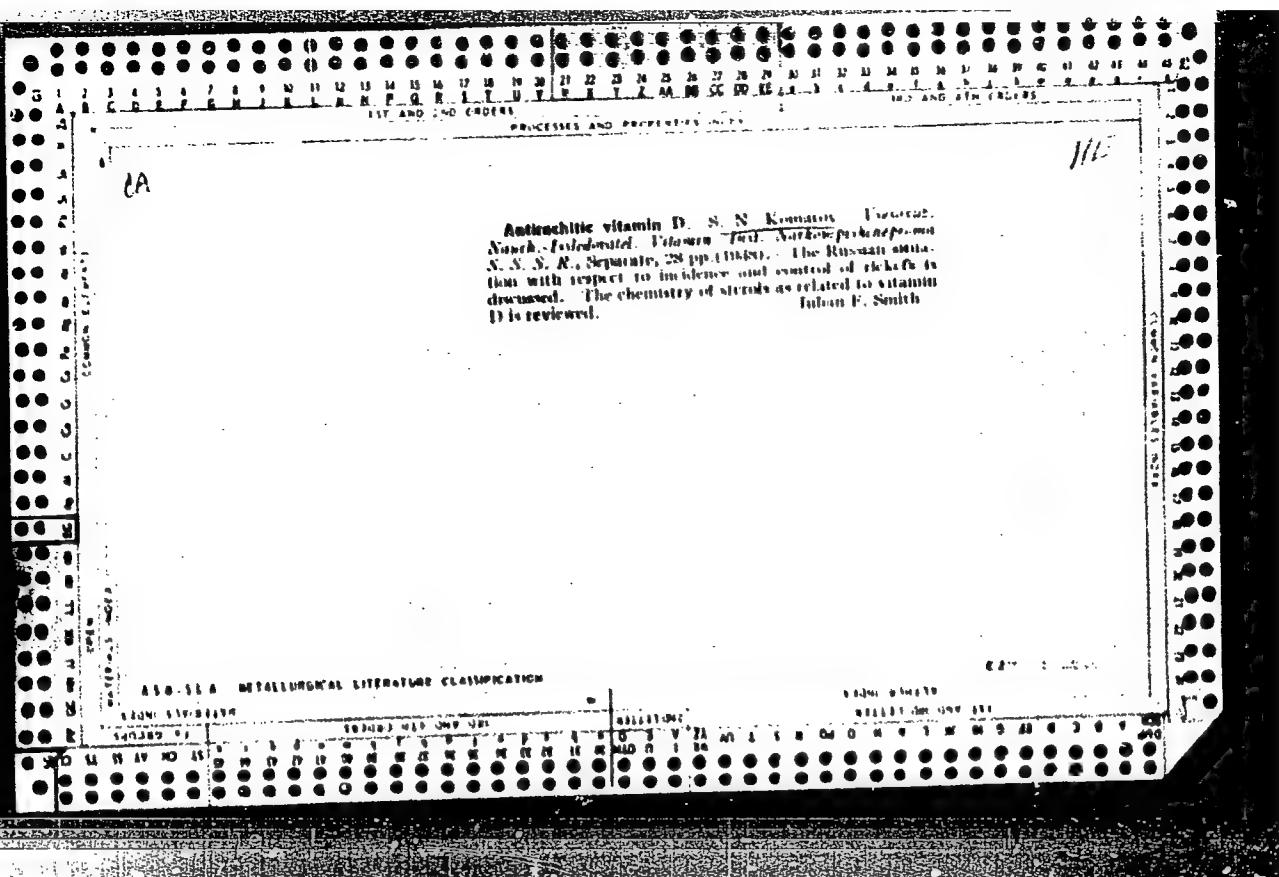
Card 2/2

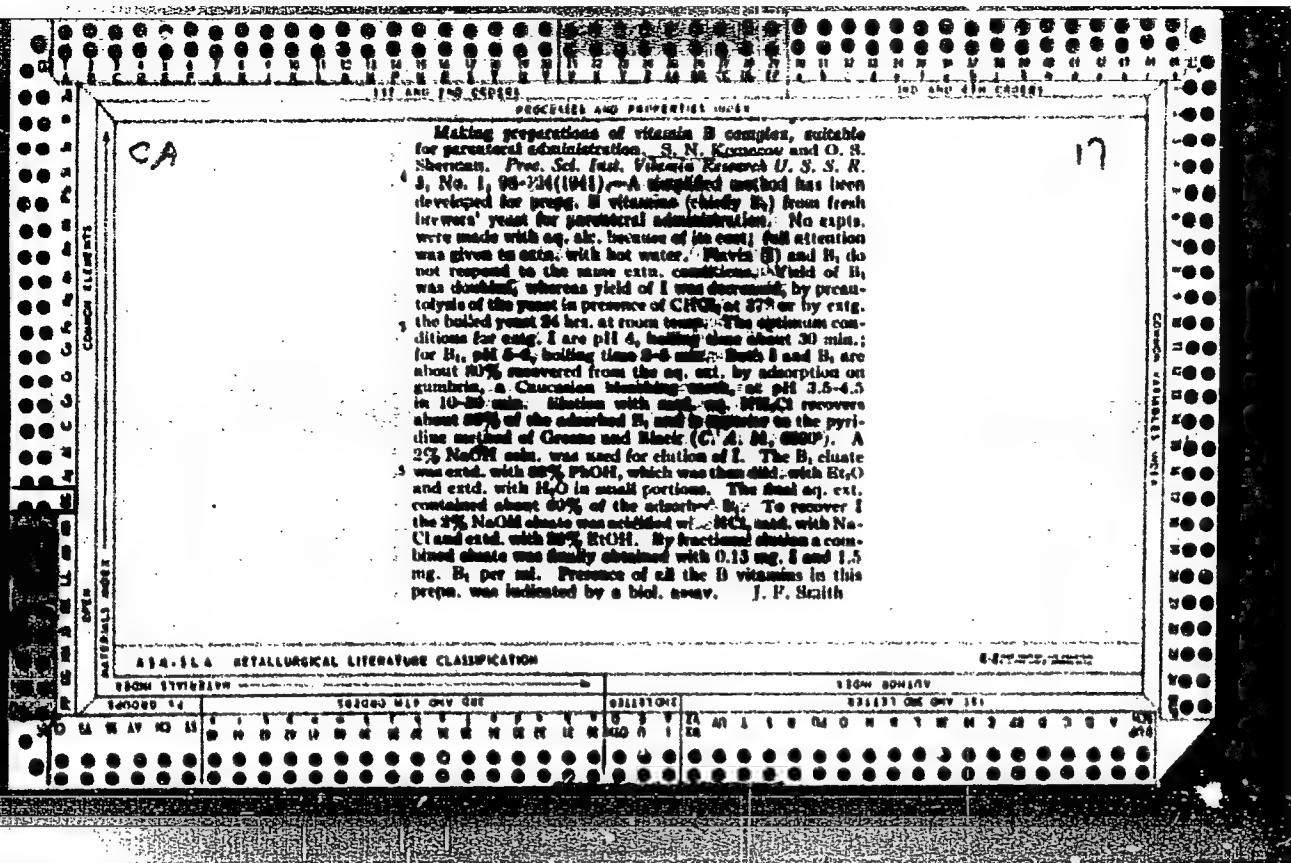
YEL'TSOV, B.V.; KOMAROV, B.M.

Using ZAUS regulators at the Novosibirsk Heat and Electric Power
Plant no.4. Priborostroenie no.5:22-23 My '64. (MIRA 17:6)

TIMOSHIN, V. S., inzh.; KOMAROV, S. M., inzh.

Adjustment of an impulse device controlling the loading of ball
mills according to a "level" pulse. Energetik 12 no.4:12-14
Ap '64. (MIRA 17:?)





Komarov, S. N.

The vitamin D content of the Baltic codfish. S. N. Komarov. *Trudy Vsesoyuz. Nauch.-Issledovatel. Vitamin.* fasc. 4, 219-111 (1953).—The liver fat of the Baltic codfish contains 80-250 I.U./ml. of vitamin D, the Murmanik codfish 10-40 I.U./ml. B. S. Levine.

Vitamins in nutrition of children: feeding flour with vitamins A and D. S. N. Kotliarov and I. N. Volkov (All-Union Sci. Research Vitamin-First, and Food Concentrate Combine, Leningrad). *Voprosy Pitanija* 13, No. 6, 32-4 (1954).—A good quality wheat flour contains only traces of vitamins A and D. For feeding children, 25-50 I.U. vitamin D and 30-35 I.U. vitamin A/g. can be mixed with the flour. The vitamins in the flour remained unchanged during 8.5 months storage under normal conditions. E. W.

KOMAROV, S.P., podpolkovnik, Geroy Sovetskogo Soyuza

When regulations are not followed. Vest. Vozd. Fl.
no. 7:64-65 Jl '60. (MIRA 13:7)
(Aeronautics—Safety measures)

KOMAROV, S., podpulkownik, bohater Zwiazku Radzieckiego; GRECZYŃ, W., kapitan, inż.

Elimination of premises causing aeronautical accidents. Wojsk
przegl 13 no.10:22-26 0 '60.

KOMAROV, Sergey Vasil'yevich; GROMOVA, V.A., red.; NAZAROVA, A.S., tekhn.
red.

[How a motion picture is produced] Kak sozdaetsia kinofil'm. Mo-
skva, Izd-vo "Znanie" Vses. ob-va po rasprostraneniuu polit. i nauchn.
znanii, 1961. 39 p. (Narodnyi universitet kul'tury. Fakul'tet lite-
ratury i iskusstva, no.6) (MIRA 14:7)

(Motion pictures—Production and direction)

KAMINSKIY, P.; KOMAROV, V.

Safe work methods. Stroitel' 8 no.9:28-29 S '62. (MIRA 15:12)
(Building—Safety measures)

KOMAROV, V.

Our aid to collective farms. Pog. delo 3 no. 4:30 Ap '57.

(MIRA 10:?)

(Firemen)

KOMAROV, V., kapitan

Improve the performance of warehouses, Voen,sviaz, 16 no.4:26
Ap '58. (MIRA 11:4)
(Warehouses)

KOMAROV, V.

Russia - Economic Policy

Stalinist program of communist construction. V. pom.profaktivu 14, no. 8, 1953.

Monthly List of Russian Accessions, Library of Congress, June 1953. Unclassified.

KOMARKOV, V.; GETMANENKO, V., starshiy master stantsii

Noninflammable cleaning solutions. Poch. delo 5 no. 7:14 Jy '59.
(MIRA 12:9)

1. Nachal'nik Novosibirskoy pozharno-ispytatel'noy stantsii
(for Komarkov)
(Cleaning compounds)

LISTOPAD, G. (Velikiy Ustyug, Vologodskaya obl.); KOMAROV, V. (Novgorodskaya obl.); FEDOROVYKH, I. (Toguchinskly rayon, Novosibirskaya obl.); SUVOROV, A. (Omsk); TROSHKOV, D. (Permskaya obl.); ZAGOROVSKIY, L.; GLOBUSOV (Sverdlovskaya obl.)

1. Readers' letters. Poch.delo 8 no.12831 D '62. (MIRA 16:1)
(Fire prevention)

KOMAROV, V.

What automation leads to under capitalism ("Automation and social process" by S. Lilly. Reviewed by V. Komarov). Sov. profsoiuzy 7 no.10:59-61 My '59. (MIRA 12:9)
(Labor laws and legislation--Dictionaries)
(Lilly, S.)

ZOTOV, I.; KOMAROV, V.

Posters are a form of concrete propaganda of leading work methods.
Sots. trud. no. 8:122-126 Ag '58. (MIRA 11:9)

1. Sekretar' partkoma metallozavoda Moskovskogo oblastnogo sovnarkhoza (for Zotov). 2. Nachal'nik otdela truda i zarabotnoy platy metallozavoda Moskovskogo oblastnogo sovnarkhoza (for Komarov). (Moscow Province--Metal industries) (Posters)

SHNEYDERMAN, M., inzh.; KOMAROV, V.

Machine for straightening wheel disks. Avt. transp. 38
no. 3:28-29 Mr '60. (MIRA 13:6)
(Machine tools)

GRAKHOVSKIY, R.; KOMAROV, V.

Heater for automobiles. Za rul. 18 no.10:24-25 0 '60.

(Automobiles—Cold weather operation) (MIRA 14:1)

KOMAROV, V., inzh.

Safety regulations for preparing mortars and concrete mixes.
Stroitel' no. 12:26-27 D '60. (MIRA 13:12)
(Industrial safety) (Mortar) (Concrete)

KOMAROV, V., inzh.

Taking measures to improve working conditions. Stroitel' no.6:25-26
Je '61. (MIRA 14:7)

(Construction industry—Safety measures)

KOMAROV, V., polirovshchik, udarnik kommunisticheskogo truda

Cool eyes, warm smile and good spirits. Obshchestv. pit.
no.11:48-49 N '61. (MIRA 15:2)

1. Zavod "Moskabel".
(Restaurants, lunchrooms, etc.)

KOMAROV, V., polirovshchik

Warmhearted woman. Obshchestv. pit. no.10:19-21 o '61. (MIRA 15:1)

1. Zavod "Moskabel".
(Moscow--Restaurants, lunchrooms, etc.—Management)

KOLCHIN, A.; KOMAROV, V., mekhanik; ARENDT, G.

Where is the new ER-10 excavator? Stroi. truboprov. 7 no.4:25
Ap '62. (MIRA 15:5)

1. Nachal'nik stroitel'nogo uchastka No.6 tresta
Soyuzprovdmekhanizatsiya (for Kolchin). 2. Nachal'nik
spetsial'nogo konstruktorskogo byuro Gazstroymashina (for
Arendt).

(Excavating machinery)

KOMAROV, V.

Guarantee certificate as an indicator of the work performed. Stroitel'
8 no.4:10-11 Ap '62. (MIRA 15:7)
(Building—Contracts and specifications)

L 12897-63

EWT(r)/ECS(g)/EWT(d)/EWT(l)/EWT(m)/BDS - AFFTC/APOC

ACCESSION NR: AP3000179

S/0029/63/000/005/0018/0018

57

56

AUTHOR: Komarov, V. (Student)

TITLE: Uniform-strength structures — the limit of possibility

SOURCE: Tekhnika molodezhi, no. 5, 1963, 18

TOPIC TAGS: uniform-strength wing, stress calculation, sandwich-type construction

ABSTRACT: V. Komarov was awarded a gold medal for his research on and calculations of a uniform-strength wing having only one point of attachment. The wing investigated had diverging longerons and the metal was so distributed as to assure a structure of maximum rigidity. The study showed that uniform-strength wings are the lightest in weight. However, their manufacture is complicated, since the sandwich-construction edges have a variable cross section. Therefore, calculations were made for a uniform-strength wing with longerons and edges having a constant cross section. Calculations showed that it was advantageous to use sectional edges. The increased rigidity of the edges decreased the stress on the long longerons and increased the stress on the short ones. The stress in a uniform-strength structure is the same in all of

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ACCESSION NR: AP3000179

its cross sections, proportional to the load. Therefore, the method of calculation is not based upon the stress diagram and is equally accurate for metals, plastics, reinforced plastics, and other materials not subject to Hooke's Law. Orig. art. has: 1 figure.

ASSOCIATION: Kuybyshevskiy aviationsionnyy institut (Kuybyshev Aviation Institute)

SUBMITTED: 00

DATE ACQ: 10Jun63

ENCL: 00

SUB CODE: AP

NO REF Sov: 000

OTHER: 000

Card 2/2

KOMAROV, V.

KOMAROV, V.

From three whales to the geoid. IUn.tekh. 2 no.1:16-21 Ja '58.
(MIRA 11:1)
(Earth--Figure)

KOMAROV, V.

Why did giant pangolins die out? IUn. tekhn. 2 no.7:41-46
J1 '58. (MIRA 11:10)
(Extinct animals) (Paleontology--Mesozoic)

KOMAROV, V.

Cosmic chemistry. IUn.tekh. 2 no.8:27-30 Ag '58.
(Cosmogony) (MIRA 12:7)

KOMAROV, V.

The 61st satellite. IUn.tekh. 3 no.1:53-54 Ja '59.
(MIRA 12:1)
(Satellites)

KOMAROV, V.

Volcano on the moon. IUn.tekh. 3 no.3:37-41 Mr '59.
(MIRA 12:4)
(Moon--Surface)

KOMAROV, V.

Solar system in 1959. IUn.tekh. 3 no.4:13-16 Ap '59.

(MIRA 12:4)

(Solar system)

KOMAROV, V., lektor

Road to outer space. Mast.ugl. 8 no.12:3-4 D '59.
(MIRA 13:4)

1. Moskovskiy planetariy.
(Space flight)

KOMAROV, V., lektor

Man strives to conquer outer space. Sov.shakht. 10 no.8:42-
44 Ag '61. (MIRA 14:8)

1. Moskovskiy planetariy.
(Astronautics)

KOMAROV, V.

The new and progressive are winning out ("Fighter planes take off" by I. Grebeniuk. Reviewed by V. Komarov. Kryl.rod. 12 no.4:19 Ap '61. (MIRA 14:7)

(Flight training)
(Grebeniuk, I.)

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S/631/61/000/002/003/013
1003/1203AUTHORS: Komarov, V., and Smirnov, M. V.TITLE: Equilibrium potentials of hafnium in mixed fluoride-chloride meltsSOURCE: Akademiya nauk SSSR. Ural'skiy filial. Institut elektrokhimii. Trudy, no. 1961, Elektrokhimiya rasplavlennykh soleykh i tverdykh elektrolitov. 19-22

TEXT: The formation of stable hafnium-fluorine complexes in the above salts can be deduced from the results of equilibrium potential measurements of hafnium and from the fact that no solid phase of any hafnium compound containing fluorine is formed when NaCl-KCl fused salts containing up to 3.4 wt % of Hf and up to 35 wt % of NaF are heated between 700 and 900°C. The equilibrium potentials of hafnium were measured against a chlorine reference electrode in NaCl-KCl fused salts containing from 0.99 to 3.4 wt % of Hf and 1.36 to 15.8 wt % of fluorine at 700, 800, and 900°C. Formulas representing the temperature dependence of the equilibrium potential and of the instability coefficient of the HfF_6^{2-} ion are given. There are 2 figures.

f

Card 1/1 REFERENCE S/631/61/002/002/013

KOMAROV, V., general-polkovnik

Acquire combat training in the field. Starsh.-serzh. no.5:10
My '62. (MIRA 15:6)

1. Zamestitel' glavnokomanduyushchego Sukhoputnymi voyskami,
nachal'nik Glavnogo upravleniya boyevoy podgotovki.
(Military education)

KOMAROV, V.

Farther and farther into the outer space. Sov.shakht. 11
no.11:41-43 N '62. (MIRA 15:11)

1. Chlen Vsesoyuznogo astronomo-geodesicheskogo obshchestva pri
Akademii nauk SSSR.
(Astronautics)

KOMAROV, Viktor

Satellites over the planet. Prir i znanie 15 no.9:18-19 N '62.

1. Nauchen komentator na Agentsiata po pechata Novosti.

KOMAROV, V.

Star map. Nauka i zhizn' 29 no.7:109 J1 '62.
(Stars--Atlases) (MIRA 16:6)

AUTHOR: Komarov, V. (Engineer, Colonel, Hero of the Soviet Union, Aviator, Cosmonaut) 60
18

TITLE: For the good of all mankind (the flight of the three-man "Voskhod" space-craft)

SOURCE: Aviatsiya i kosmonavtika, no. 12, 1964, 27-30

TOPIC TAGS: "Voskhod" series, astronaut training, manned satellite, spacecraft, spacecraft instrumentation, weightlessness, space flight, space observa-
tion, "Voskhod" spacecraft

ABSTRACT: The author was one of the three-man crew aboard the Soviet "Veskhod" spaceship which was launched on 12 October 1964 and landed, after 10 orbits around the Earth, on 13 October 1964. In the present article, he discusses some of the aspects of that historic space flight. The presentation is in a popular and non-technical style and is aimed at the non-specialist reader. The other members of the crew (K. P. Feoktistov, the scientist, and B. R. Yegorov, the astronaut-medic) are identified and their duties are briefly defined. The flight program, which the author states was scheduled for execution in a single 24-hour period and which was carried out completely, included the following basic goals: 1) the testing

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ACCESSION NR: AP5001808

of the design and operational characteristics of the new multi-place piloted spacecraft, its systems and its instrumentation; 2) the study of the capability for work and cooperation in flight of a group of astronauts, consisting of specialists in various fields; 3) the conducting of scientific investigations of a physical-technical and medical-biological nature during the conditions of an extended space flight; 4) the continued study of the effect of various spaceflight-related factors on the human organism. The need for a close spirit of cooperation and friendly mutual assistance among all 3 crew members for the successful execution of this flight program is noted. Mention is made of the training of the crew needed for actual launching, and attention is called to the fact that each was an actual part in the creation of the "Voskhod" vehicle, and of its systems and instrumentation. The author states that the so-called "medical-biological preparation" (that is, testing and training) was carried out according to an abbreviated program". This program apparently included vestibular, G-force-support and emotional training elements. Noting that all six "Vostok" flights were single-man flights, the author points out that the expanded 3-man crew of the "Voskhod" spacecraft made it possible not only to enlarge the scope of the scientific program of the undertaking considerably, but also to place the observations on a more scientific and higher level. The need to suspend observa-

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ACCESSION NR: AP5001808

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tions for the time required by the astronaut for sleep, for example, which was inherent in the "Vostok" flights, was easily eliminated in the case of the "Voskhod" by the simple expedient of conducting the observations in shifts. Greater objectivity through a diversity of interpretation is also afforded by a crew rather than by a single astronaut. Among the other items mentioned by the author one might single out the fact that for the first time the astronauts were not encumbered by the usual "space-suit", the fact that the "soft-landing" system performed perfectly, the fact that a "new principle for the control of the spacecraft was experimentally checked out" (this "new principle" is not further discussed or identified), and the fact that a series of experiments with liquids and gases under the conditions of weightlessness were carried out. Orig. art. has: 3 photographs.

ASSOCIATION: None

SUBMITTED: 00

ENCL: 00

SUB CODE: SV, PH

NO REF SOV: 000

OTHER: 000

Card 3/3 pyp

ACC NR: AP7005431

SOURCE CODE: UR/0209/66/000/009/0040/0043

AUTHOR: Komarov, V. (Colonel; Cosmonaut; Hero of the Soviet Union)

ORG: none

TITLE: Scientific expeditions in space

SOURCE: Aviatsiya i kosmonavtika, no. 9, 1966, 40-43

TOPIC TAGS: astronaut, spaceborne earth observation, aurora, space flight

ABSTRACT:

Cosmonaut V. Komarov quotes his fellow cosmonaut K. P. Feoktistov who made observations from space: "Observations of the horizon were made for obtaining data on the clarity of the boundary of the horizon for the purpose of selecting a reference layer in the optical range for ensuring navigation and orientation in orbital and interplanetary flights when it is necessary to use the earth as a reference celestial body during astronavigational measurements and for orientation of spaceships and automatic space vehicles. In most cases on the daytime side of the earth the horizon is observed both as the boundary of the atmosphere and the earth and a "layer" of a blue aureole with a clear upper boundary. The upper boundary of this aureole is clearer than the apparent boundary between the earth and atmosphere. After the ship entered the earth's shadow it was possible to observe a layer of brightness at an altitude of 60-100 km above the boundary between the earth and atmosphere. The brightness of the layer was close to the brightness observed at the horizon of the earth, illuminated by the moon. The crew

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ACC NR: AP7005431

was most impressed by an aurora which was observed in Antarctica prior to emergence from the shadow. The picture was as follows: horizon, dark sky, upper layer of brightness illuminated by the moon, and above it -- rays perpendicular to the horizon with an altitude of 6-8° and intervals of about 2°. Along the horizon the aurora occupied the entire visible field of view. It was found possible to measure the altitude of stars above the visible horizon, which in the future will make it possible in space flights to automatically determine spacecraft position and compute its trajectory of motion and necessary corrections. It was possible to observe luminescent particles in the portholes of the ship when the direction of observation was perpendicular to the sun's rays. Presumably those were dust particles separated from the ship, illuminated by the sun and situated several meters from the ship." The objective of this article is to demonstrate that the role of the scientific specialist aboard a spacecraft is exceptionally varied and important for direct observations in space. A table accompanying the text lists different types of possible scientific flights, the recommended orbital altitude, optimum number of crew members and duration of flight and the equipment which should be carried and the experiments to be carried out (however, it is noted that this information was taken from the foreign press). Orig. art. has: 1 table. [JPRS: 38,677]

SUB CODE: 22 / SUBM DATE: none

KOMAROV, V. A.

MEN, S.A., dots.; KOMAROV, V.A., red.; REGICHVA, M.N., tekhn.red.

[Conveying machines and installations] Transportiruiushchie mashiny
i ustanovki. Moskva, Izd-vo M-va technogo flota SSSR, 1951. 503 p.
(Conveying machinery) (MIDA 11:2)

VOROBTSOV, Yevgeniy Stefanovich; KOMAROV, V.A., retsenzent; ANDREYeva,
L.S., red.; TIKHONOVA, Ye.A., tekhn.red.

[Mechanization of transfer operations of hold, freight car, and
warehouse materials in seaports] Mekhanizatsiia triumykh,
vagonnykh i skladskikh peregruzochnykh rabot v morskikh portakh.
Moskva, Izd-vo "Morskoi transport," 1961. 346 p.

(MIRA 15:5)

(Materials handling) (Harbors)

Komarov, V. A.

AID P - 4069

Subject : USSR/Power

Card 1/1 Pub. 26 - 27/33

Author : Komarov, V. A., Eng.

Title : Defects in preassembled current transformers.

Periodical : Elek. sta., 12, 55, 1955

Abstract : The article describes defects found in transformers delivered at a new power plant's construction site. Gaging transformers were shipped back to the factory.

Institution : None

Submitted : No date

KOMAROV, V.A.

KOMAROV, V.A., insh.

Rope carrying system in repairing 400 kv cutouts. Energetik 5
no.10:19-21 0⁰57. (MIRA 10:12)
(Electric lines—Maintenance and repair)

BAZHINOV, A.G., podpolkovnik meditsinskoy sluzhby; KAMORSKIY, N.M., podpolkovnik;
KOMAROV, V.A., podpolkovnik, kand.khimicheskikh nauk

New substances and methods for disinfecting hospital rooms (as
revealed by foreign studies). Voen.-med. zhur. no.7:53-56 Jl '61.

(MIRA 15:1)

(DISINFECTION AND DISINFECTANTS) (HOSPITALS--SANITATION)

KOMAROV, V.A.

Ecology of penduline titmouse (*Remiz pendulinus caspius* Poelzam)
in the Volga Delta. Trudy Astr. zap. no.5:262-268 '61.

(MIRA 16:8)

(Volga Delta—Titmice)

BOROV, V. P.; ZHEREBTSOVA, K. I.; KRASNOV, L. V.; KOMAROV, V. A.; LITVIN, V. F.;
NEFEDOV, Yu. A.

"Investigations of the Reactions of Type (d,p) on Isotopes of Zn, Ni,
and Fe⁵⁸."

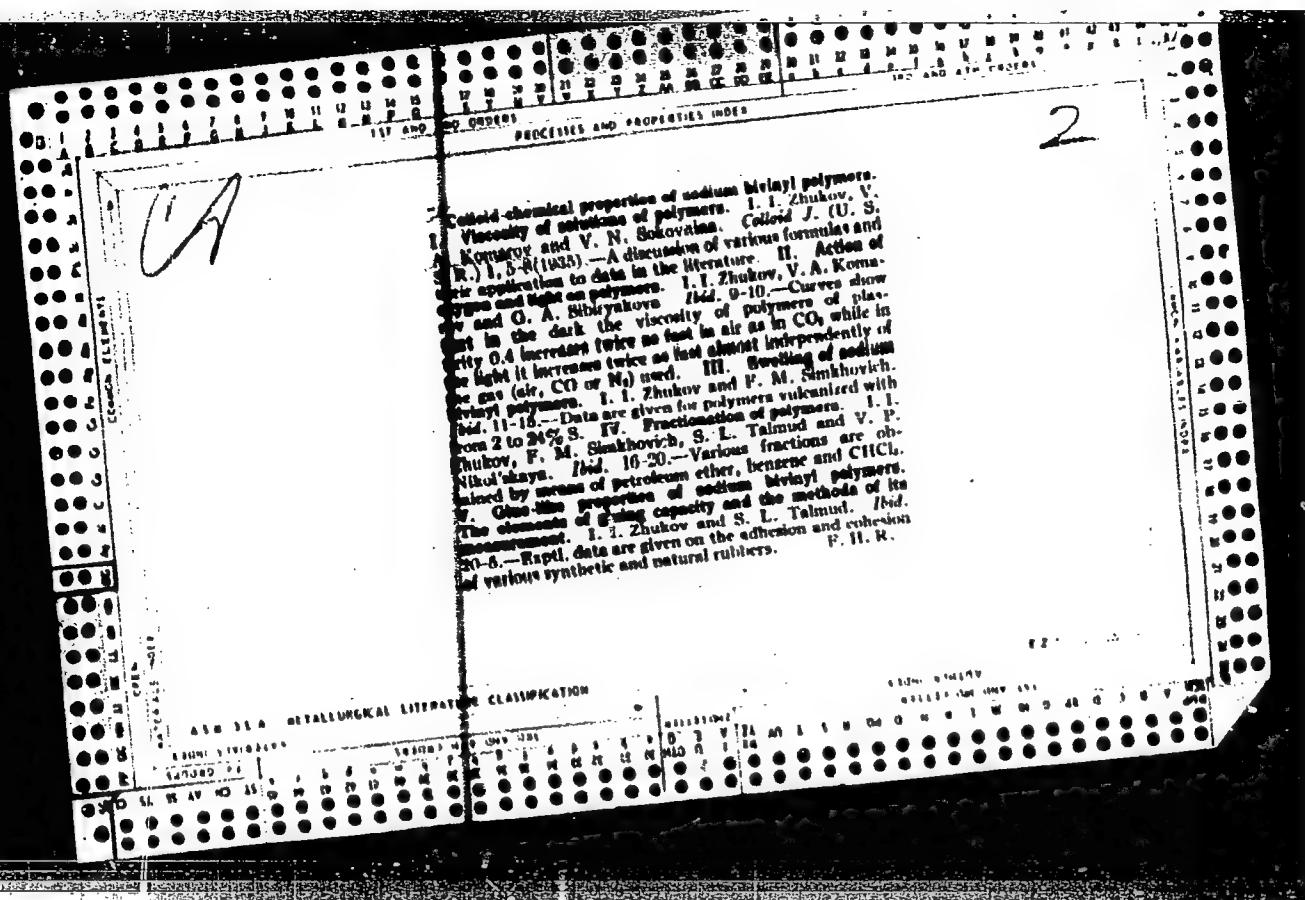
report submitted for All-Union Conf on Nuclear Spectroscopy, Tbilisi, 14-22
Feb 64.

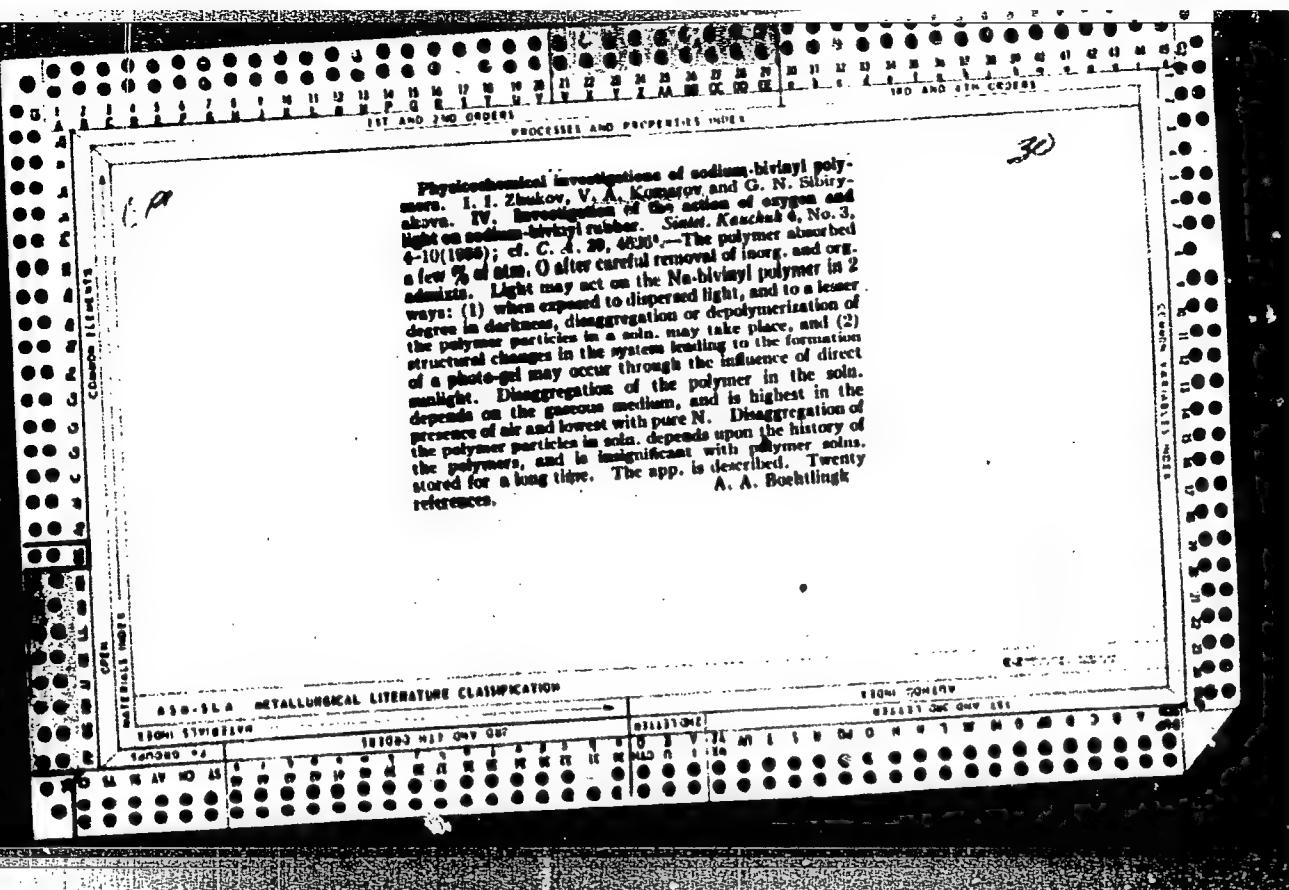
Radiyevyy Institut (Radium Inst)

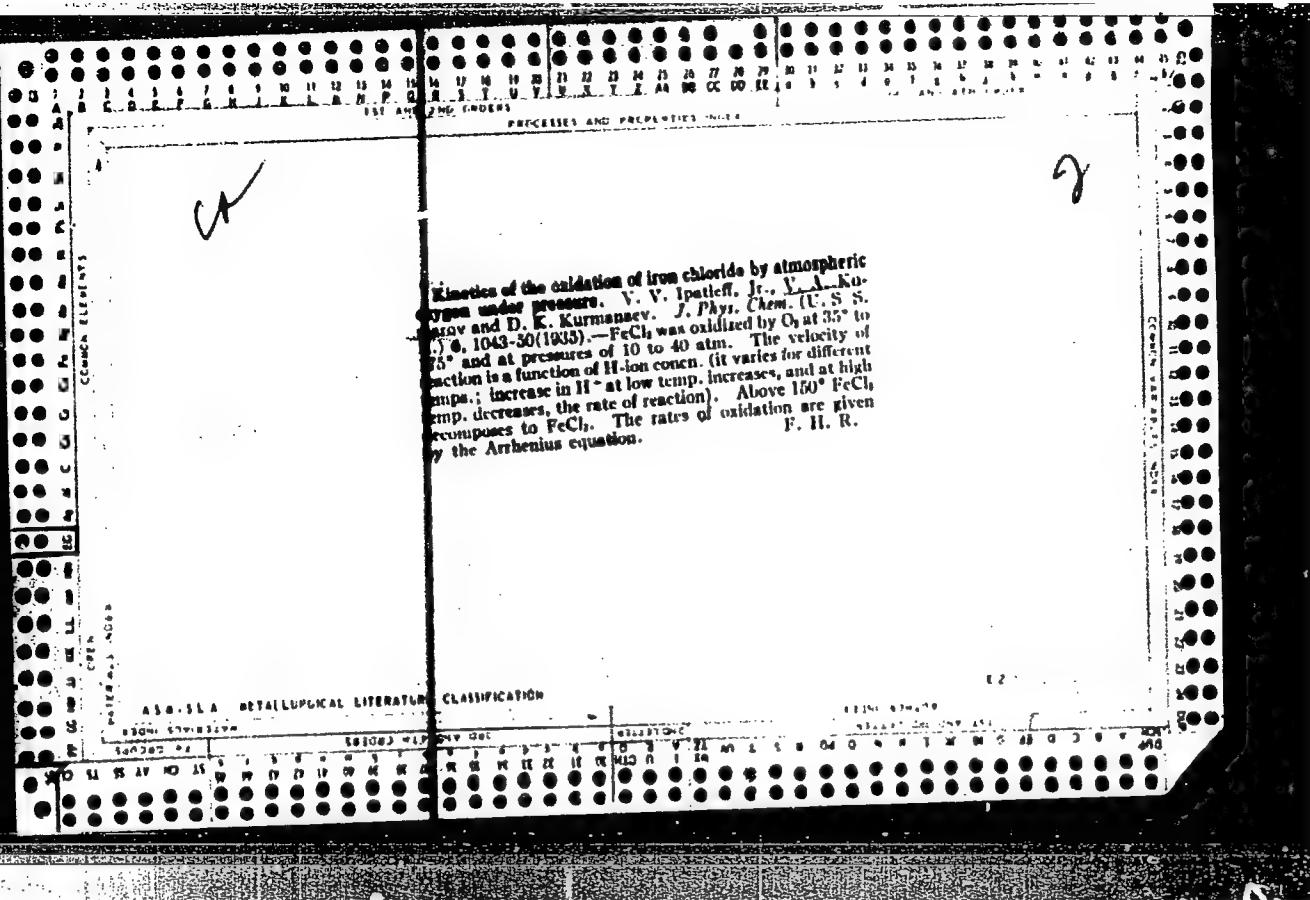
KOMAROV, V.A.; MUSIYACHENKO, T.I.

Let's overcome our shortcoming. Zemledelie 26 no.12:20-23 D '64.
(MIRA 18:4)

1. Direktor sovkhoza "Yarul'skiy", Rybinskogo proizvodstvennogo
upravleniya, Krasnoyarskogo kraya (for Komarov). 2. Glavnyy agronom
sovkhzoza "Yarul'skiy", Rybinskogo proizvodstvennogo upravleniya,
Krasnoyarskogo kraya (for Musiyachenko).







The influence of heating on sodium bromide rubber in solution. I. I. Zhukov, V. A. Komarov, S. F. Valter and R. I. Gribova. *Sintet. Kaučuk* 1936, No. 2, 4-9. — Na butadiene rubber (I) was purified by pptn. with alc. from C_6H_6 soln. and dried in ratios to evntr. wt. C_6H_6 soln. of purified I of 0.1, 0.2, 0.3, 0.8 and 1.8% concn. were prep'd., and 15-cc. samples were put in glass ampoules, sealed and heated at 100°, 115°, 130° and 150°, resp., for different times. The changes were followed in 2 ways: (1) by viscosity changes in an Ostwald viscometer at 25° in N; the relative viscosity η_r (the ratio of the time of flowing of the soln. to the time of flowing of pure solvent) was detd. and by using η_r , the mol. wt. of the polymer in soln. (M_r) and the sp. vol. of polymer in soln. (σ) were calc'd.; and (2) by measuring the surface tension between a C_6H_6 soln. of rubber and water (σ) (the surface tension showed the degree of oxidation). Heating up to 100° slightly changed η_r , M_r and σ , but from 100° to 150°, η_r , M_r and σ decreased rapidly, and σ nearly approached the viscosity of the solvent. The time of heating had the same effect upon η_r , M_r and σ , the values of which decreased to a certain min., after which further heating increased them. The 0.5% soln. of I in several org. solvents were heated for 24 hrs at 130° and η_r , M_r and σ were detd. It was found that the nature of the solvent has a decided influence on the degree of depolymerization.

(the ratio of the mol. wts. after and before heating of the polymer M_1/M). For 1-pinen, M_1/M was 0.4; *d*-pinene 0.29; PhCl 0.10; C_6H_6 0.11; C_6H_5Cl 0.07; dichloroethylene 0.23; cumene 0.33. To 100 cc. of 0.5% soln. of polymer in C_6H_6 was added 0.01 g. of antioxidant (Agerite, Neozone, Albaran, etc.), the mixts. were sealed in ampoules and heated for 24 hrs. at 130°. These tests were made in air, *N* and *in vacuo*. All antioxidants prevented decrease in viscosity and surface tension on heating. With decrease in the antioxidants from 0.5 to 0.0150%, their stabilizing action diminished. In the absence of an antioxidant *in vacuo*, depolymerization was much less, and *in vacuo*. The heating of the polymer with an antioxidant *in vacuo* resulted in less depolymerization than without antioxidant (e.g. without antioxidant 28.0, with, 20.2; M 37,500 and 52,700; f 60.6 and 77.3; v 1.01 and 2.29). Therefore, the antioxidant exerted its stabilizing action upon the polymer, even when δ was practically absent. The action of the antioxidant consisted: (1) in preventing oxidation of the polymer, and (2) in a general stabilizing effect upon high-mol. substances. The authors suggest that these antioxidants be designated: "stabilizers" or "antiagers." Seventeen references are included. A. Postoff

A. Period

ASB-52A METALLURGICAL LITERATURE CLASSIFICATION

APPROVED FOR RELEASE: 06/13/2000

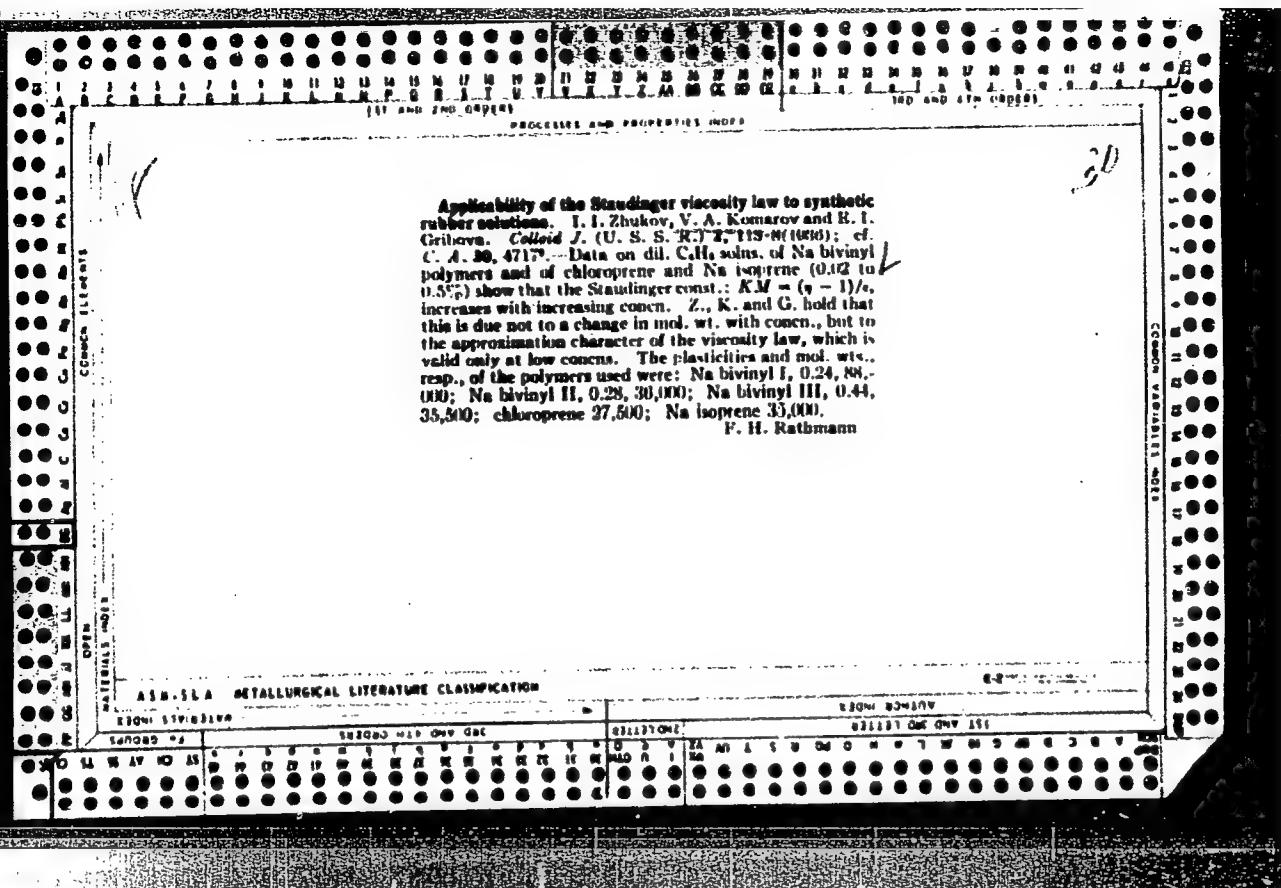
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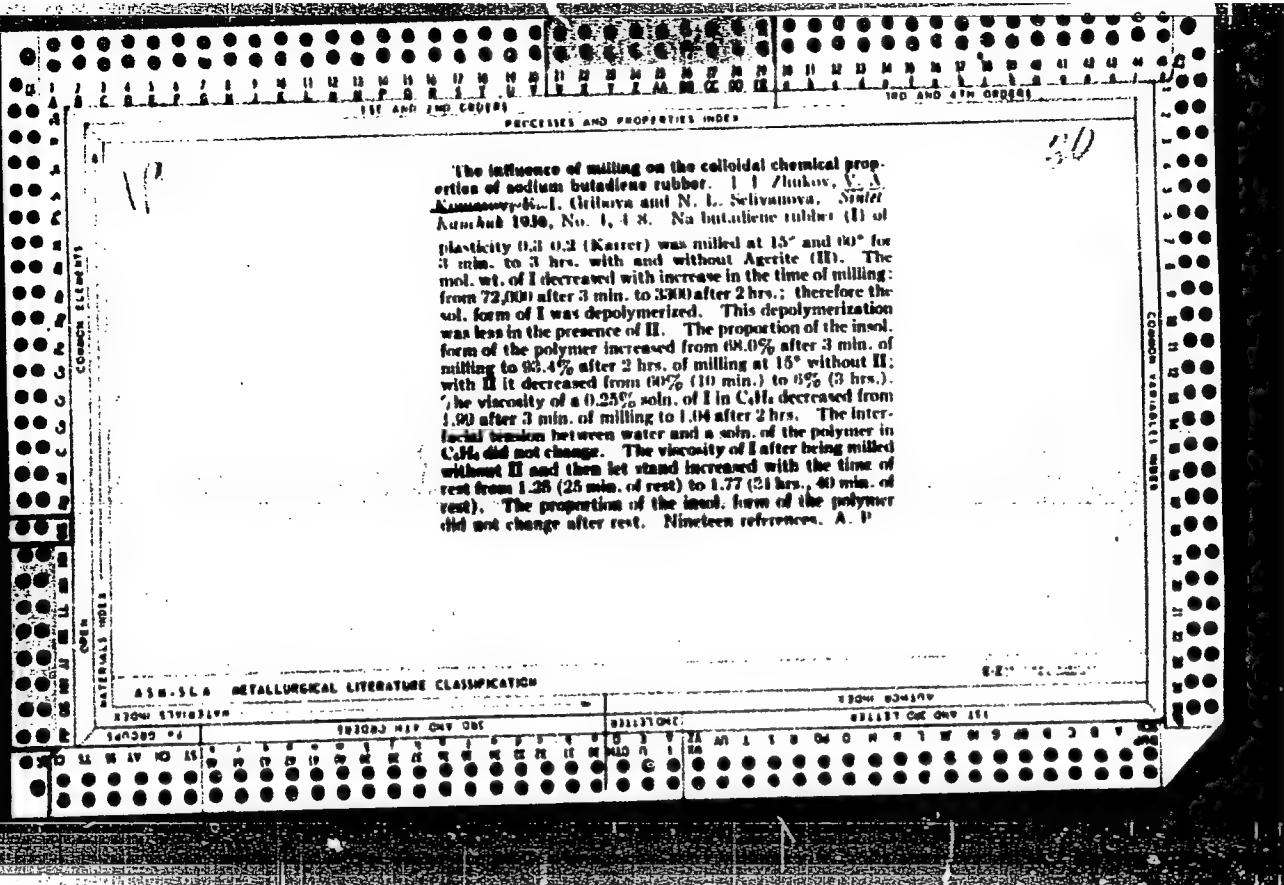
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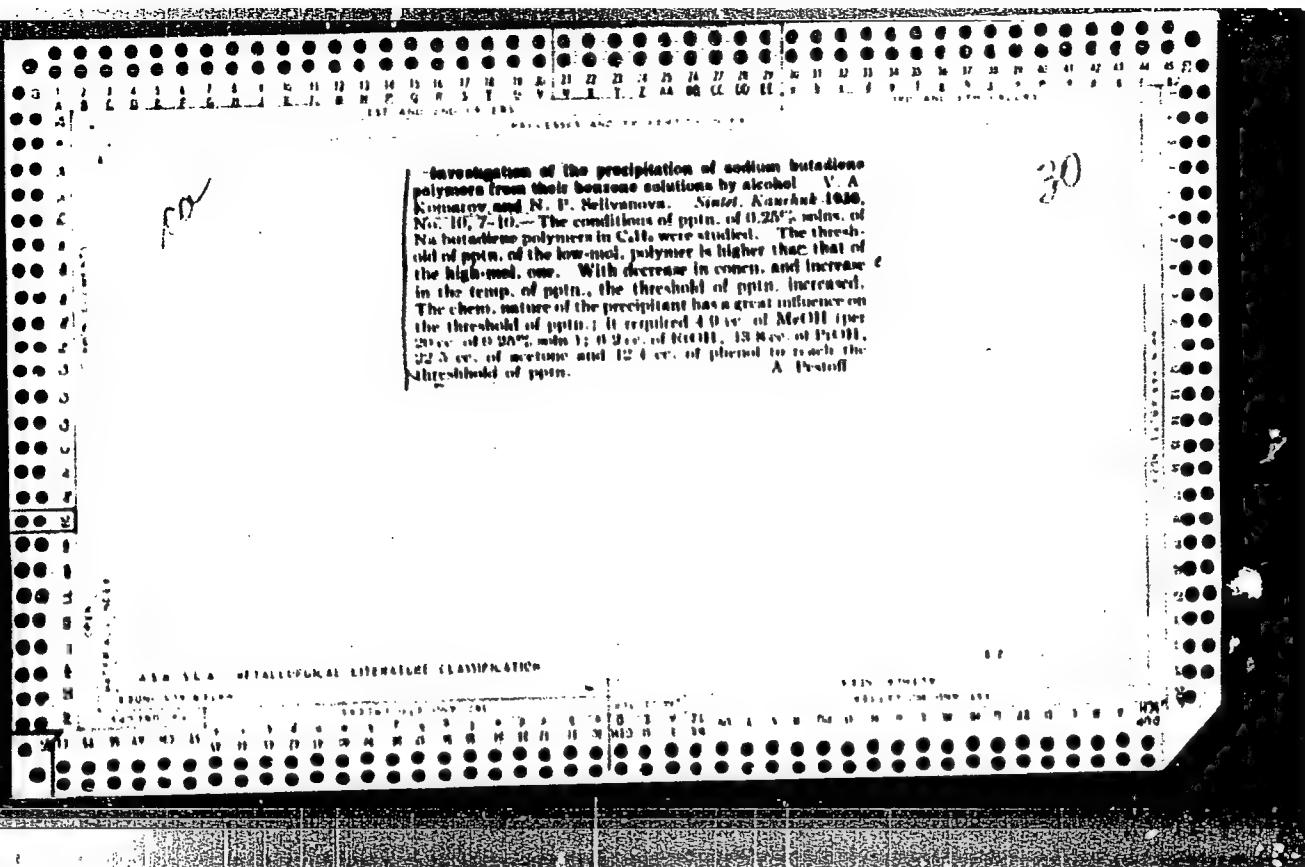
30

The application of the viscosity law of Staudinger to
synthetic-rubber solutions. I. I. Zhukov, V. A. Komarov,
and E. I. Grilova. *Sintet. Kaučuk* 1936, No. 2, p. 12.
The viscosities of 0.5-0.025% solns. of Na-butadiene
polymers in C_6H_6 were detd. For solns. with viscosity
lower than 1.30, Staudinger's viscosity const. diminished
with decrease in the concn. The errors in deg. mol. wts.
by viscometric data were ruled to be not over 18%.
A. Pestoff
Eleven references.

ASIS SLA METACONTRAL LITERATURE CLASSIFICATION



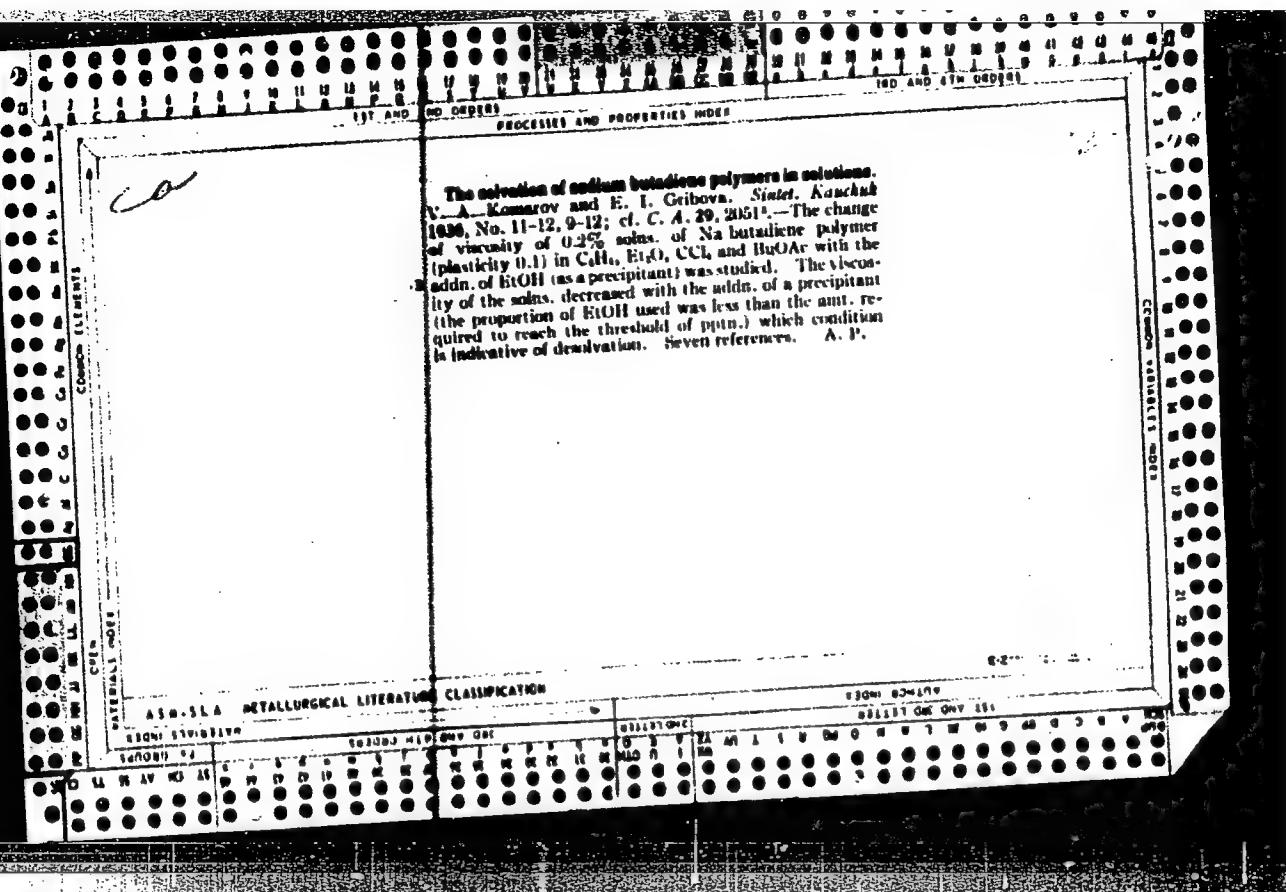


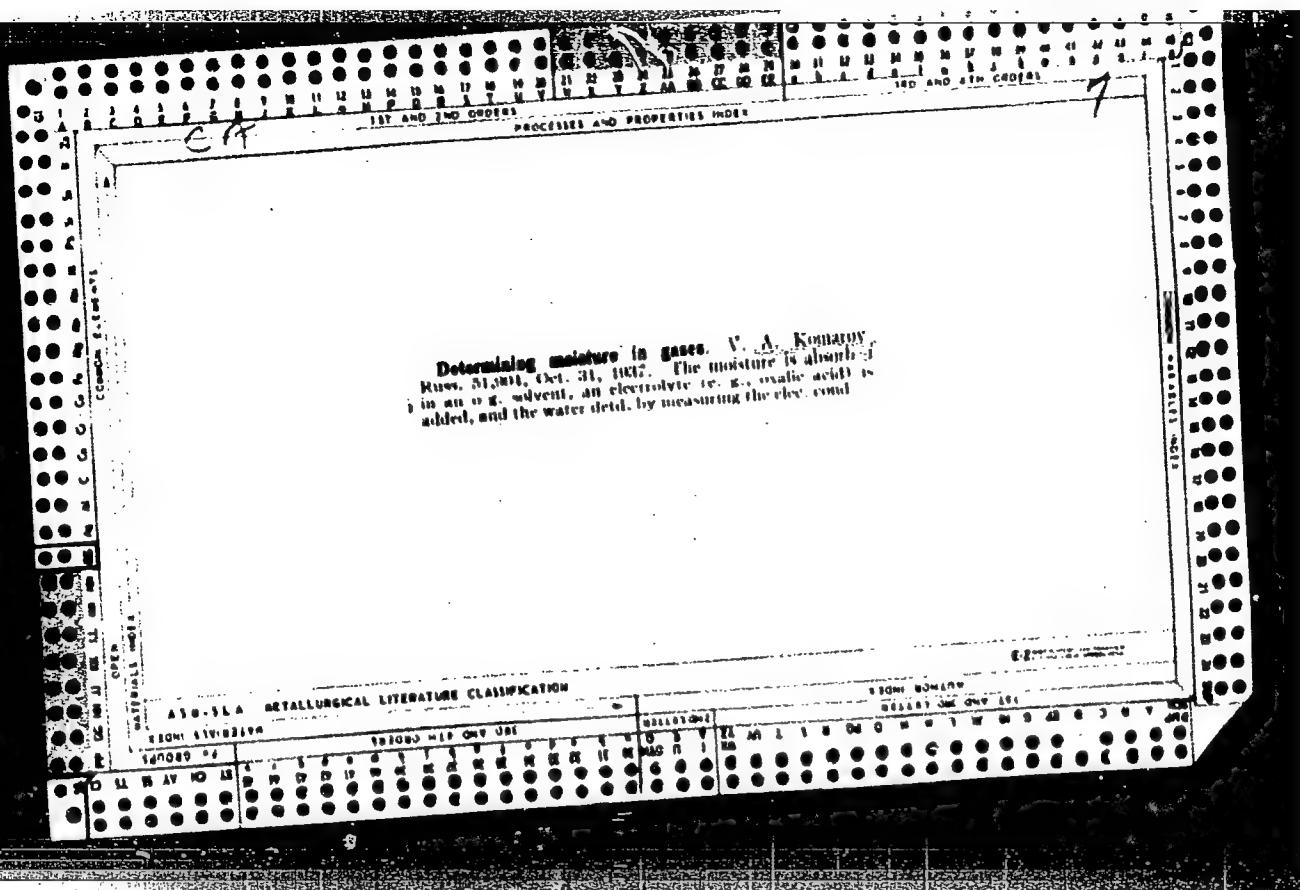


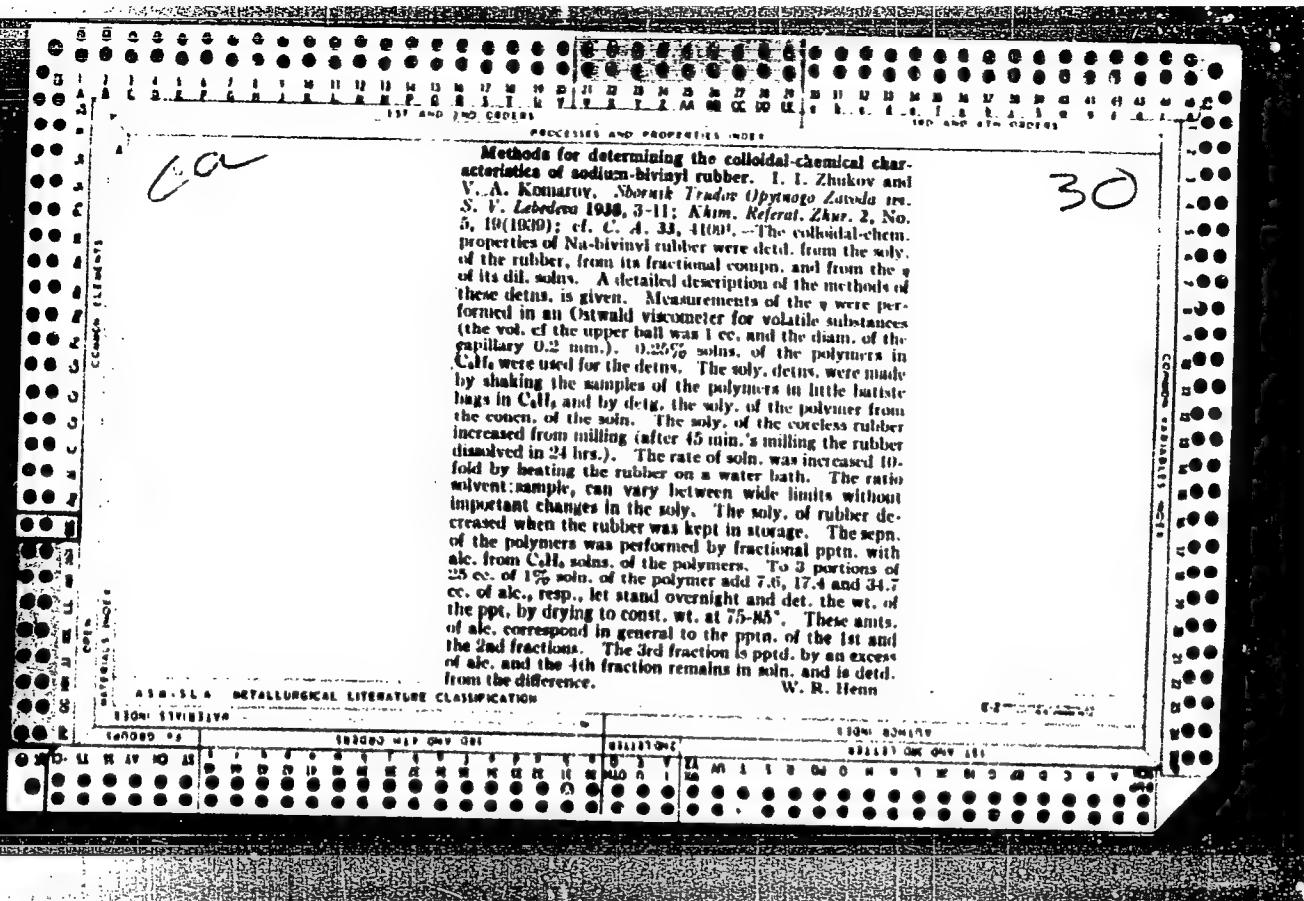
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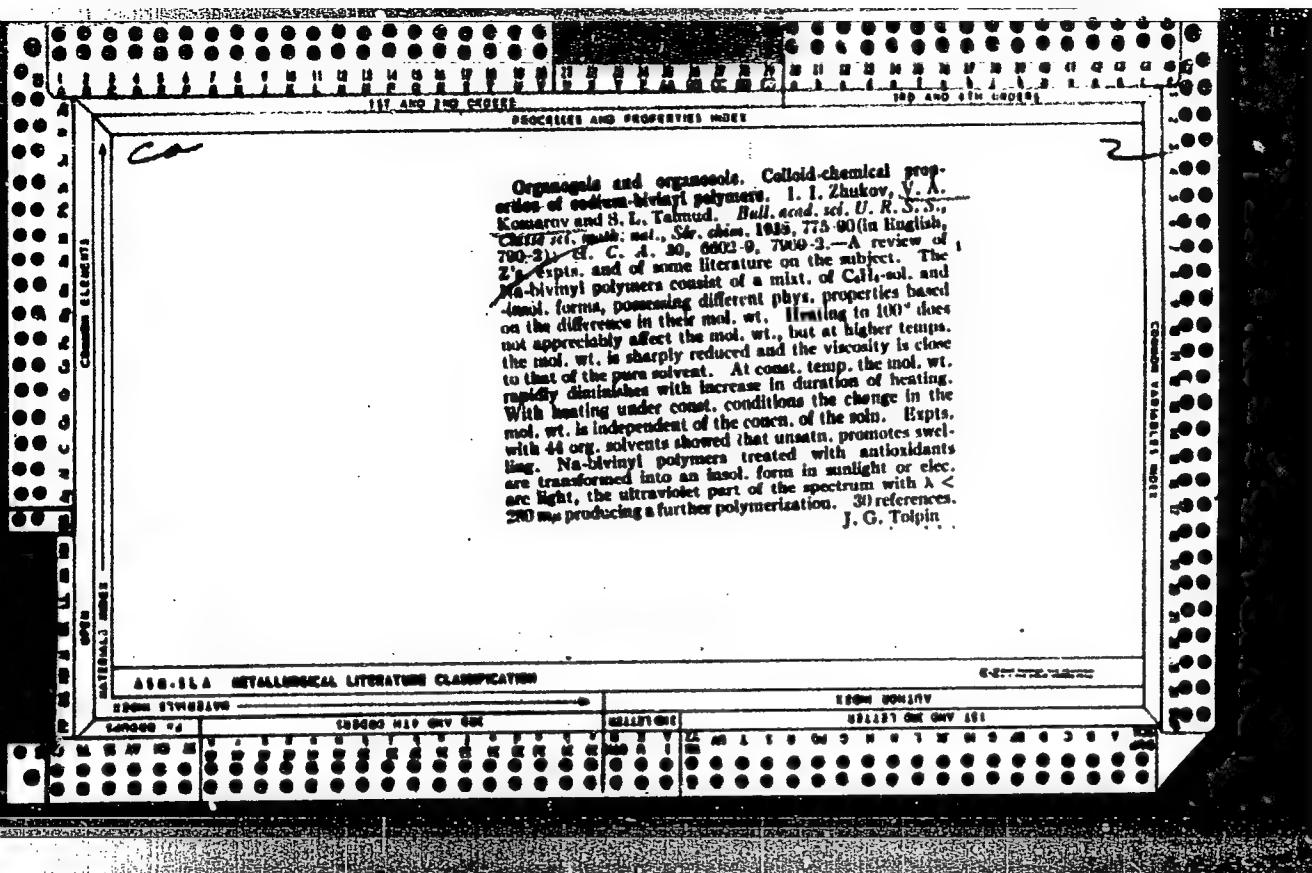
30

Determination of the molecular weight of sodium butadiene polymer by the method of partial vulcanization. V. A. Krasarov and S. F. Val'ter. *Sintet. Kaučuk* 1956, No. 10, 11-13; *ref. C. A.* 53, 4384. The polymer was dissolved in CaH_2 and the solvent was removed *in vacuo* at room temp. A mixt. of 10 g. of polymer and 1.57 g. of tetrakis(methyldiethoxy)disulfide was vulcanized at 100° for 1 hr. and again dissolved in CaH_2 . The dissolved part of the polymer was fractionated by addn., in a separatory funnel, of RIOH till the first flocculent ppt. was formed; this ppt. was washed with RIOH , dried, weighed and analyzed for total S. RIOH was added to the soln. remaining from the 1st ppt. until a ppt. was just formed; the ppt. was collected and so on, until 4 fractions were taken as a unit (1 atom); the 2nd fraction was found to be 1.6 atoms; the 3rd, 2 atoms; the 4th, 0 atoms; the mol. wts. of the fractions were 13,200, 13,000, 12,700 and 1,000, or approx. 0.25 of the mol. wt. of natural rubber. A. Pestov









KOMAROV, V.A.

✓New trends in the scientific research work of the Chemical Institute of the Leningrad University, V. A. Komarov, Period Leningrad Univ. 2, No. 1, 201-210(?) - A discussion of the work which is planned for the coming 6-year period. Special emphasis is placed on the study of alloys and the production of synthetic (Fischer-Tropsch) fuels.
I. Koytar Leach

CA

2

Surface determination by the method of adsorption of report. V. A. Komarov, V. M. Drodova, and E. A. Chernikova. Zhur. Fiz. Khim. 23, 1141-51 (1949).—Adsorption of N_2 by MgO , ZnO , CdO , and Cr_2O_3 (all ppd. from nitrate salts, with NH_4 , and heated to 800°), by Cr_2O_3 from $(NH_4)_2Cr_2O_7$, by a natural and 3 artificial samples of SiO_2 , and by metallic Mg and Zn was detd. at -183° . The surface S_1 calc'd. from the adsorption at which the linear portion of the isotherm starts usually was smaller than that, S_2 , calc'd. from the Brunauer-Emmett-Teller theory (C.A. 33, 40377), the greatest difference being 60%, whereas surface S_3 calc'd. from the Harkins-Jura equation (C.A. 34, 51265) was, on the av., = 0.78 S_1 . In several instances, the latter equation was applicable only if its const. had 2 different values at small and great relative pressures. Adsorption of butane at 0° was detd. for ZnO , MgO , the 4 SiO_2 samples, and one of the Cr_2O_3 samples. Here also S_3 usually was greater than either S_1 or S_2 . If the areas occupied by a mol. of N_2 and butane are 16.3 and 28 A^2 , resp., the surface accessible to butane after long adsorption was in all instances smaller than that accessible to N_2 ; the ratio was, e.g., 0.17-0.48 for the SiO_2 samples and 0.21, 0.39, and 0.73 for ZnO , Cr_2O_3 , and MgO . The const. of butane adsorbed within a few min. were smaller than those after long adsorption. Apparently, there are 3 fractions of the total surface: (a) rapidly and (b) slowly accessible to butane and (c) accessible only to N_2 . Heating in $EtOH$ at 320 - 440° for hrs. lowered S_1 of MgO and ZnO and increased S_1 of CdO on heating, of Mg, and Zn.

J. J. Bikerman

KOMAROV, V. A., DROZDOVA, V. M., SHIF, G. A.

Reduction, Chemical

Determination of the starting temperature of reduction of metallic oxides with hydrogen. Uch.zap. Len.un. No. 150, 1951.

9. Monthly List of Russian Accessions, Library of Congress, November 1952 ~~1953~~, Uncl.

1. KOMAROV, V. A.
2. USSR (600)
4. Chemical React on - Velocity
7. Connection between initial temperatures of reactions and variations of free energy
for monotype reactions, Dokl. AN SSSR 87, No. 4, 1952.
9. Monthly List of Russian Accessions, Library of Congress, February 1953. Unclassified.

grid. Univ. No. 6949c. The relation between ΔF_{10}° and T_R was studied. According to literature data

$T_{AF=0}$ calcd. from previous reaction *Ibid.* 36-10. The relation between T_R and $T_{AF=0}$ was studied. According to literature data there is an approx. linear relation between T_R and $T_{AF=0}$, as well as between T_R and $T_{AF=0}$. III. Reduction of metal oxides with alcohols. *Ibid.* 41-8. T_R , ΔF_{10}° , and $T_{AF=0}$ were studied for the reactions $\text{EtOH} + \text{MO} = \text{C}_2\text{H}_5\text{O} + \text{M} + \text{H}_2\text{O}$ where $\text{M} = \text{H}_2\text{O}$ and $\text{EtOH} + \text{MO} = \text{C}_2\text{H}_5\text{O} + \text{M} + \text{H}_2\text{O}$ where MO is CuO , Cu_2O , CaO , ZnO , Cr_2O_3 , TiO_2 , MnO_2 , and MgO . T_R was read from the appearance of gas evolution and also with an indicator (HCl) soln. of NaOH . T_R , ΔF_{10}° , and $T_{AF=0}$ were calcd. from calculated data. For the 2nd of the above reactions there was a linear relation between T_R and $T_{AF=0}$ as well as for T_R and $T_{AF=0}$. For the 1st reaction the relation was expressed by a rising curve. *Rejent. Zhar. Khim.* 1954, No. 27-53 53-51. M. No. 10

KOMAROV, V.A.

(2)

Relation between initial temperatures of reactions and changes of free energy for reactions similar in kind. V. A. Komarov (A. A. Zhdanov State Univ., Leningrad). *Zhur. Fiz. Khim.* 27, 1748-50 (1953); cf. Tsvetkov, *Trudy Inst. Geol. Nauk, Akad. Nauk S.S.R.* 106, 67 (1949).—The initial (T_i) and equil. (T_e) temps. of reaction, ΔH_{rxn}° , and ΔF_{rxn}° were calcd. from literature data for the reduction by H of the oxides of the bivalent metals Cu, Zn, Cd, Ni, Fe, Co, Pb, and Mn and for the thermal decompn. of the carbonates of Ca, Mg, Fe, Zn, Ba, Pb, and Sr. The values of T_i and T_e for the reduction by EtOH and iso-PrOH of MgO, MnO, TiO₂, Cr₂O₃, ZnO, CdO, Fe₂O₃, CuO, and Cu₂O were measured; values of ΔF_{rxn}° were calcd. The value of T_i is a nearly linear function of ΔF_{rxn}° for these reactions. Data are tabulated and graphed. The linear relation between ΔF_{rxn}° and T_i is predicted in a theoretical discussion. J. W. Lowenberg Jr.

KOMAROV, V.A.

Connection between the initial reaction temperature and the variation of free energy. Part 1. Reactions of the reduction of metal oxides by hydrogen. Uch.zap.Len.un.169:29-35 '53.
(MLRA 9:6)
(Oxides) (Reduction, Chemical) (Chemical reaction, Heat of)

KOMAROV, V.A.

Connection between the initial reaction temperature and the
variation of free energy. Part 2. Carbonate dissociation
reactions. Uch.zap.Len.un.169:36-40 '53. (MLRA 9:6)
(Carbonates) (Dissociation) (Thermal analysis)

KOMAROV, V.A.

Connection between the initial reaction temperature and the variation of free energy. Part 3. Reduction of metal oxides by alcohols. Uch.zap.Len.un.169:41-48 '53. (MIRA 9:6)
(Oxides) (Reduction, Chemical)

KOMAROV, V.A.

USSR/ Chemistry - Physical chemistry

Card 1/2 Pub. 147 - 15/21

Authors : Komarov, V. A., and Chernikova, Ye. A.

Title : Effect of certain hydroxide admixtures on the dehydration of Al(OH)_3

Periodical : Zhur. fiz. khim. 29/10, 1876-1882, Oct 1955

Abstract : The process of Al(OH)_3 dehydration (pure aluminum hydroxide and Al(OH)_3 containing admixtures of other hydroxides) was investigated by the continuous oven suspension and thermographic methods. A strong effect of other hydroxide admixtures on the dehydration of Al(OH)_3 was definitely established. Foreign hydroxide admixtures result in the reduction in the

Institution : Leningrad University im. A. A. Zhdanov, Inst. of Chem.

Submitted : March 19, 1955

Card 2/2 Pub. 147 - 15/21

Periodical : Zhur. fiz. khim. 29/10, 1876-1882, Oct 1955

Abstract : number of endothermal maxima corresponding to the maximum rate of water separation and in the reduction of the temperature interval between the maxima. The specific surface of Al(OH)_3 compounds containing admixtures was found to be smaller than the surface of pure compounds. Nine references: 6 USSR and 3 Germ. (1924-1954). Tables; graphs.

Komarov, V. A.

Passage of 13-PrOH over Mn was examined, up to 400° . On

an intermediate, the mol. wt. of the alc. from EtOH , $\text{C}_2\text{H}_5\text{OH}$, CH_3OH , and $\text{C}_2\text{H}_5\text{OH}$. The final product of action is an Mn oxide complex, $\text{MnO}_2 \cdot \text{MnO}_4$, which retains the x-ray a

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CIA-RDP86-00513R000824110002-3"

Reaction of metallic oxides with alcohols. I. A. N. S. and
dium oxides isopropyl alcohol. I. A. N. S. and
P. Tomichev (State Univ., Leningrad). *Zh. fiz. ch.* 1957, 31, 26.
Form 26. 24.8.40 (USSR). 1. A. N. S. and
P. Tomichev over VdKh M-14. 0.1 g. of
a mixture of the oxides in the ratio 1:1.5 is
placed in a tube containing 10 ml. of
isopropyl alcohol. The tube is heated in
water (these conditions) for 10-15 min. The
reaction is accompanied by a brightening
and is apparently due to the formation of
dehydrogenating catalyst. The ratio of isopropyl
alcohol formed in the reaction to the
volume of the reaction mixture is
about 10. The reaction is
repeated in the same way
with the addition of 0.1 g. of
catalyst. The reaction
is accompanied by a brightening
which is accompanied by lesser stability
Reaction begins to be evident on VdKh only at about 200°
G. M. Kosolapov

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CIA-RDP86-00513R000824110002-3"

KOMAROV, V. A.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000824110002-3"

Dietrich *4Eg/j/4E2c(j)*

Interaction of *1* with *2* in the presence of copper
oxides and isopropyl alcohol.

1. 51.3% CuO, 31.8% Cu₂O, 16.9% CuO₂ + Cu₂O
Working reaction: 10-PrOH + 3 CuO \Rightarrow Cu₂CO + Cu₂O
+ H₂O; 10-PrOH + Cu₂O \Rightarrow Cu₂CO + 2Cu + H₂O
After 10 min and reaction the starting materials 1 + 2
have disappeared. The reaction mixture is a
yellow-green colour.

2. By catalyst *2* in isopropyl alcohol. *Bad. 200-11.*
II. Iron oxides and isopropyl alcohol. *Bad. 200-11.*
The interaction of gaseous 10-PrOH (1) with the following
oxides to be applied in catalytic (1),
10-PrOH + CuO₂ + Cu₂O + CuO + Cu₃O₂ + Cu₂CO + Cu₂CO₃ + Cu₂CO₄ + Cu₂CO₅ + Cu₂CO₆ + Cu₂CO₇ + Cu₂CO₈ + Cu₂CO₉ + Cu₂CO₁₀ + Cu₂CO₁₁ + Cu₂CO₁₂ + Cu₂CO₁₃ + Cu₂CO₁₄ + Cu₂CO₁₅ + Cu₂CO₁₆ + Cu₂CO₁₇ + Cu₂CO₁₈ + Cu₂CO₁₉ + Cu₂CO₂₀ + Cu₂CO₂₁ + Cu₂CO₂₂ + Cu₂CO₂₃ + Cu₂CO₂₄ + Cu₂CO₂₅ + Cu₂CO₂₆ + Cu₂CO₂₇ + Cu₂CO₂₈ + Cu₂CO₂₉ + Cu₂CO₃₀ + Cu₂CO₃₁ + Cu₂CO₃₂ + Cu₂CO₃₃ + Cu₂CO₃₄ + Cu₂CO₃₅ + Cu₂CO₃₆ + Cu₂CO₃₇ + Cu₂CO₃₈ + Cu₂CO₃₉ + Cu₂CO₄₀ + Cu₂CO₄₁ + Cu₂CO₄₂ + Cu₂CO₄₃ + Cu₂CO₄₄ + Cu₂CO₄₅ + Cu₂CO₄₆ + Cu₂CO₄₇ + Cu₂CO₄₈ + Cu₂CO₄₉ + Cu₂CO₅₀ + Cu₂CO₅₁ + Cu₂CO₅₂ + Cu₂CO₅₃ + Cu₂CO₅₄ + Cu₂CO₅₅ + Cu₂CO₅₆ + Cu₂CO₅₇ + Cu₂CO₅₈ + Cu₂CO₅₉ + Cu₂CO₆₀ + Cu₂CO₆₁ + Cu₂CO₆₂ + Cu₂CO₆₃ + Cu₂CO₆₄ + Cu₂CO₆₅ + Cu₂CO₆₆ + Cu₂CO₆₇ + Cu₂CO₆₈ + Cu₂CO₆₉ + Cu₂CO₇₀ + Cu₂CO₇₁ + Cu₂CO₇₂ + Cu₂CO₇₃ + Cu₂CO₇₄ + Cu₂CO₇₅ + Cu₂CO₇₆ + Cu₂CO₇₇ + Cu₂CO₇₈ + Cu₂CO₇₉ + Cu₂CO₈₀ + Cu₂CO₈₁ + Cu₂CO₈₂ + Cu₂CO₈₃ + Cu₂CO₈₄ + Cu₂CO₈₅ + Cu₂CO₈₆ + Cu₂CO₈₇ + Cu₂CO₈₈ + Cu₂CO₈₉ + Cu₂CO₉₀ + Cu₂CO₉₁ + Cu₂CO₉₂ + Cu₂CO₉₃ + Cu₂CO₉₄ + Cu₂CO₉₅ + Cu₂CO₉₆ + Cu₂CO₉₇ + Cu₂CO₉₈ + 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Bolotov, Boris Aleksandrovich, Vyacheslav Aleksandrovich Komarov,
and Tat'yana Vsevolodovna Nizovkina

Prakticheskiye raboty po organicheskому katalizu (Practical
Studies in Organic Catalysis) [Leningrad] Izd-vo Leningr.
univ., 1959. 194 p. Errata slip inserted. 4,120 copies printed.

Sponsoring Agency: Leningrad. Universitet. Redaktsionno-izdatel'skiy
soviet.

Resp. Ed.: B. N. Dolgov, Professor; Ed.: Ye. V. Shchemeleva;
Tech. Ed.: Ye. G. Zhukova.

PURPOSE: This book is intended for the personnel of scientific
research institutes and factory laboratories. It will be of
interest to teachers and students of advanced courses in
chemistry and chemical technology vuzes. It may also be used
as a manual to aid in setting up and performing various

Card 1/6

Practical Studies in Organic Catalysis

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operations with catalytic methods, and in organizing effective work practices.

COVERAGE: The book describes the principal apparatus used to produce catalytic reactions at normal and higher pressures, methods of producing and studying catalysts, and the methods of producing those catalytic reactions which embrace the main branches of organic catalysis. The authors thank K. P. Katkova, I. M. Stroyman, Ye. A. Chernikova, N. P. Usacheva, and R. M. Adrov. References accompany each chapter.

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